Research on the Synthesis Process of Isopropyl Xanthogen Polysulfide

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Abstract: The synthesis process of isopropyl xanthogen polysulfide was investigated in this study, and the influence of the raw material ratio, reaction temperature and reaction time of each reaction stage on the production yield was discussed. Based on the experimental results the optimum synthesis conditions for each stage were obtained. In the synthesis of sodium isopropyl xanthate, toluene was used as solvent, the molar ratio of isopropanol, carbon disulfide and NaOH was 1.00 : 0.95 : 0.90, the reaction temperature was in the range of $10 \sim 15$ °C, and the reaction time was $12 \sim 14$ h. In the synthesis of isopropyl xanthate ester, the molar ratio of sodium isopropyl xanthate and hydrogen peroxide was 1.00 : 0.60, the hydrogen peroxide was added during a period of $5 \sim 6$ h, the reaction temperature was maintained at $25 \sim 30$ °C, and the reaction was completed about $50 \sim 60$ min after the addition of hydrogen peroxide. In the synthesis of isopropyl xanthate polysulfide, the molar ratio of isopropyl xanthate ester and sulfur was 1.02.0 and the reaction temperature was $80 \sim 90$ °C.

Keywords: sodium isopropyl xanthate; isopropyl xanthate ester; isopropyl xanthogen polysulfide; synthesis process



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